

Click Chemistry Approaches for the Synthesis and Functionalization of Macromolecules

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Abstract: Click chemistry methods are powerful tools in the hands of synthetic chemists. Cu(I)-catalyzed azide-alkyne [3+2] dipolar cycloaddition reaction (CuAAC) is accepted as “the cream of the crop” of click chemistry techniques due to its orthogonality and near-perfect yields in very mild conditions. Besides CuAAC, Diels–Alder (DA), thiol–ene, atom transfer nitroxide radical coupling (ATNRC), and aldehyde–aminoxy reactions are other effective ligation methods in terms of chemoselectivity and reaction yields at moderate conditions, thus they are also accepted as click chemistry reactions. Click chemistry and controlled polymerization techniques have been utilized extensively in the synthesis of well-defined and elegant macromolecular structures.

Keywords: Click chemistry, Functionalization of macromolecules, Orthogonal reactions.

1. INTRODUCTION

The introduction of new technologies has been demanding intelligently developed polymers with improved physical and chemical properties, such as tailor-made molecular structures and functionalities, well-defined molecular weight, and narrow molecular weight distribution. Various polymerization and ligation methods have been employed to tackle this challenge. Cu(I)-catalyzed 1,3-dipolar azide-alkyne cycloaddition reaction (CuAAC) [1], developed by Sharpless based on Huisgen dipolar cycloaddition reaction [2], is one of the most effective and widely used ligation methods in the synthesis of macromolecular compounds [3-7]. Therefore, nowadays this reaction is named as “Sharpless-type click reaction”.

According to Sharpless, synthesis of reliable modular blocks with powerful and selective functional units is the key step for the efficient preparation of any molecules with desired chemical and structural properties [1]. This notion laid the foundations of “click chemistry” approach. There are some strict requirements for reactions to be regarded as a click chemistry technique. The reactions must be modular, tolerant to diverse functional groups other than the targeted one, require no or benign solvents, have high conversion rates at moderate conditions with high regioselectivity, and need simple and nonchromatographic purification methods [8].

The definition of “click chemistry” approach was later revised by Barner-Kowollik *et al.* [9]. According to the revised definition, stoichiometric amounts of the functional building modules should be attached in an acceptable time-scale with high conversion rates and the targeted product should be removed from the reaction mixture via simple large-scale purification methods.

Besides CuAAC, atom transfer nitroxide radical coupling (ATNRC), Diels–Alder (DA) cycloaddition, thiol–ene, thiol–yne, and aldehyde–aminoxy click reactions have been employed together with controlled polymerization (CRP) methods in the preparation of various tailor-made polymers with elegant molecular structures.

2. SHARPLESS-TYPE CLICK REACTION (CuAAC)

The Huisgen cycloaddition reactions between alkyne and azide moieties are conducted at relatively high thermal conditions, yielding 1,4- and 1,5-triazoles [10]. Later, Sharpless and Medlar independently developed this synthetic method and reported that the presence of Cu(I) salt provides this reaction with exclusive regioselectivity to produce 1,4-disubstituted 1,2,3-triazoles (Figure 1) [11, 12]. Besides, CuAAC requires very mild reaction conditions and mostly performed at ambient temperature [13-16]. Due to the orthogonality and nearly perfect chemoselectivity of CuAAC, it is accepted as the “cream of crop” of the click reactions [17] and thus, there are copious amount of literature work reporting the use of CuAAC in the preparation of well-defined polymers with various topologies, including linear [18-22], star-shaped [23,

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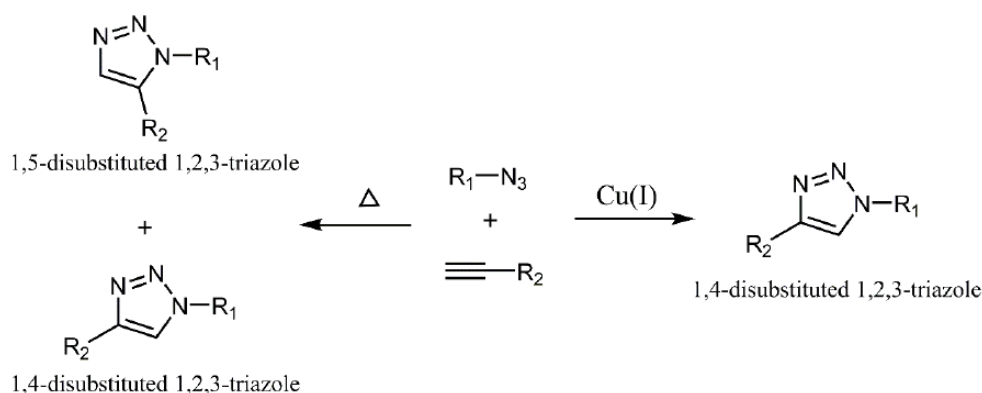


Figure 1: Huisgen (on the left) and Sharpless (on the right) type 1,3-dipolar azide-alkyne cycloaddition reactions.

24], miktoarm star [25, 26], graft [27-30], cyclic [31, 32], hyperbranched [33-35], and crosslinked [36] polymers.

In the reaction mechanism of CuAAC, proposed by Sharpless [37], coordination of Cu(I) to alkynes leads to copper-acetylide intermediate and then followed by the attachment of azide units, yielding a six membered Cu-containing metallacycle which is another intermediate species. Finally, the latter intermediate is transformed into triazole ring rapidly. The obtained 5-membered heteroatom triazole ring has high chemical stability and its large dipole moment (5D) imparts it hydrogen bond forming capacity [38, 39].

Atom transfer radical polymerization (ATRP) is one of the most used CRP method along with CuAAC conjugation technique. This polymerization technique attaches halogen atom to the end-groups of the produced polymer. Then, these alkyl-halide end-groups are easily converted into alkyl-azide units [40]. Various functional groups of monomers, small- or macromolecules can be transformed into azide or alkyne groups via suitable chemical modification techniques. Besides, due to chemical and thermal stability of azide or alkyne groups, various polymerization initiator molecules with these functionalities were used in the preparation of linear, star, and graft polymers via CRP techniques, such as ATRP [41], reversible addition-fragmentation chain transfer polymerization (RAFT) [42-44], ring opening polymerization (ROP) [42, 44-47], and nitroxide-mediated polymerization (NMP) [48].

The introduction of a small molecular compound with suitable functional units to a macromolecular structure is a relatively straightforward synthetic route and can be achieved via various ligation methods, including click chemistry methods [49]. However, combination of two or more macromolecules and

isolation of the targeted polymer through simple large scale purification methods is a quite challenging procedure. Fortunately, CuAAC is an excellent combination method, even in the case of equimolar amount of the polymers with azide and alkyne functional polymers, to give exceptionally high conversion rates. In a typical example, PMMA-PSt-PtBA and PMMA-PSt-PEG miktoarm star terpolymers were prepared through CuAAC and CRP methods. Poly(MMA-*b*-St) having an alkyne functional moiety was synthesized by ATRP of methylmethacrylate (MMA) and NMP of styrene (St) employing an initiator molecule with alkyne, bromide and alkoxyamine functional groups. Subsequently, an equimolar amount of azide functional poly(*tert*-butylacrylate) (PtBA) was treated with alkyne functional Poly(MMA-*b*-St) copolymer through CuAAC reaction, yielding PMMA-PSt-PtBA miktoarm star terpolymer. Whereas, excess amount of azide functional poly(ethylene glycol) (PEG) was reacted with Poly(MMA-*b*-St) in the same manner, and then the crude product was precipitated in methanol to remove the unreacted PEG. The monomodal GPC curves indicated that only the targeted polymers were obtained, as well as the success of the performed CuAAC reactions [50].

The orthogonality (tolerance towards various functional groups other than the targeted one) of a ligation method becomes extremely important in one-pot reaction procedures. The most important advantage of the one-pot reaction procedures is that the purification methods are applied only to the targeted final polymers, not to the intermediates [51]. Zhang *et al.* reported tailor-made PSt-PCL-PDMA and PEO-PCL-PDMA miktoarm star polymers in a one-pot procedure [52]. The core molecule with alkyne, bromide and hydroxyl functional groups was attached to azide functional PSt (or PEG) via CuAAC reaction, yielding a macroinitiator with bromide and hydroxyl

functional groups for simultaneous ATRP of 2-(dimethylamino)ethyl methacrylate (DMA) and ROP of ϵ -caprolactone (ϵ -CL), respectively. In this one-pot mechanism, the use of equimolar amount of core molecule and PSt (or PEG) is vitally important to produce well-define miktoarm star polymer.

The synthesis of block copolymers can be accomplished using living polymerization methods, such as ionic polymerization (cationic or anionic), ATRP, RAFT, and NMP. This approach necessitates to perform consecutive polymerization reactions. A single living polymerization method or combinations of them can be used to produce block copolymers, depending on the end-functional units of the precursor polymers. The polymer obtained from the first reaction is employed as a macroinitiator for another monomer in the next step to extend the polymer chain. This approach limits the type of monomers to be used in the preparation of block copolymers [53]. The polymerization initiator functional groups at the termini of the precursor chains dictate the type of the monomers of the next polymer blocks. Whereas macromolecular structures with azide and alkyne end-functional units can be efficiently attached to each other via CuAAC at quantitative and near-perfect conversion rates. Opsteen *et al.* [54] elegantly prepared linear diblock PMMA-*b*-PEG, PSt-*b*-PEG, PMMA-*b*-PSt, and triblock PEG-*b*-PSt-*b*-PEG via Cu(I)-catalyzed click reaction between alkyne and azide groups. GPC chromatograms of the synthesized block copolymers were observed to be monomodal. In another example, Quémener *et al.* [55] reported diblock PSt-*b*-PVAc copolymer through CuAAC of azide functional poly(vinyl acetate) (PVAc) and alkyne functional PSt which were obtained by RAFT polymerization of respective monomers. He *et al.* prepared diblock PEG-*b*-PSt copolymer via ATRP of St using bromide end-functional PEG as the macroinitiator. Following azidification of the bromide end groups of PEG-*b*-PSt, alkyne functional PCL was

attached to the diblock copolymer via CuAAC reaction, yielding triblock PEG-*b*-PSt-*b*-PCL copolymer [56].

Cu(I)-catalyzed 1,3-dipolar cycloaddition reactions were also used in the synthesis of various macromolecular structures in a step-growth manner [57]. Zhou *et al.* reported the synthesis of polytriazoleimides by CuAAC step-growth polymerization between dialkyne and diazide functional monomers [58]. Besset *et al.* prepared starch-based polytriazoles through CuAAC reactions between diazide and dialkyne functional dianhydrohexitol monomers [59].

3. DIELS–ALDER CLICK REACTIONS

Diels-Alder (DA) reaction mechanism was invented by two Nobel Laureates (1950), Otto Diels and Kurt Alder [60]. This reaction is a straightforward [4+2] cycloaddition reaction between a conjugated diene and dienophile, giving a stable cyclohexene moiety (Figure 2) [61]. Conjugated dienes and dienophiles have 4 π - and 2 π -electron systems, respectively. The temperature requirements for these reactions depend on the selection of diene-dienophile pairs [4]. For example, cycloaddition of anthracene and maleimide occurs at relatively high temperatures around 110 °C, whereas furan-maleimide cycloaddition reactions can be conducted at relatively low temperatures. But the products obtained from the latter reaction is not thermally stable, retro-DA reaction of the furan-maleimide adduct is observed to take place above 60 °C in solutions [62]. Thermal reversibility of this reaction makes it suitable for the preparation of thermo-responsive materials [63,64]. Hetero Diels-Alder reactions between dithioester functional group and butadiene can be conducted around 50 °C [65], whereas cyclopentadiene can be coupled to dithioester unit at ambient temperatures. The adducts obtained from the reaction of dithioesters and cyclopentadiene are thermally stable up to 80 °C [4].

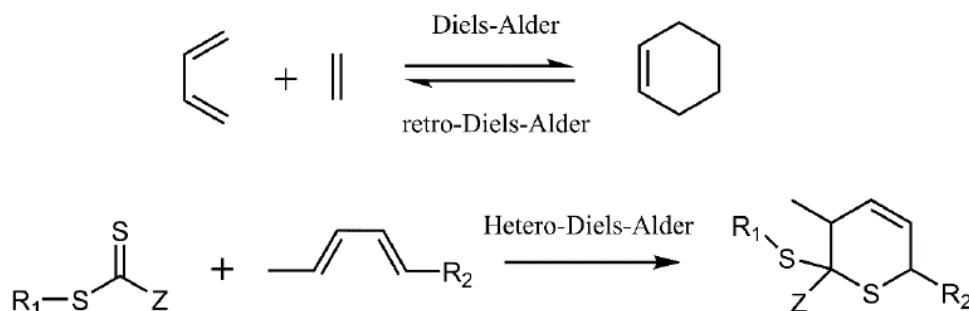


Figure 2: Diels-Alder, retro-Diels-Alder, and hetero-Diels-Alder reactions.

DA reactions do not require transition metal catalyst and have been employed in the synthesis of elegant macromolecular structures including homopolymers, dendrimers, star polymers, block and graft polymers. Durmaz *et al.* prepared linear triblock PEG-*b*-PSt-PMMA and PCL-*b*-PSt-*b*-PMMA copolymers via in situ (one-pot) CuAAC and Diels-Alder [4+2] reactions in an elegant way, demonstrating the orthogonality of these ligation mechanisms [51]. In some cases, maleimide group is protected with furan to prevent any side reactions and furan protecting unit can be removed by rising the temperature to over 100 °C in suitable reaction conditions. Tunca and Hizal group synthesized PEG and Jeffamine macrostructures with furan-protected maleimide groups at both ends and then removal of furan unit was carried out by refluxing in toluene. The maleimide end functional groups of the obtained polymers were reacted with anthracene moiety of a trifunctional compound. Then, nitroxy and bromide functional groups of the trifunctional compound were used as the initiators in consecutive NMP of St and ATRP of tert-BA, respectively, yielding heteroarm H-shaped terpolymers [66].

Bousquet *et al.* reported preparation of an elegant graft copolymer through RAFT polymerization and hetero-DA conjugation techniques. Firstly, random copolymer of St and *trans*, *trans*-2,4-hexadienylacrylate (ttHA) monomers was synthesized. Then, homopolymer of n-BA was prepared via RAFT polymerization and attached to P(St-*r*-ttHA) copolymer via hetero-DA click method, yielding P(St-*g*-nBA) graft copolymer [67].

4. THIOL-ENE CLICK REACTIONS

Thiol-ene reaction is a hydrothiolation reaction of C=C double bond and has click-type characteristics (Figure 3). Thiol-ene reactions are facile and versatile processes, can be performed under variety of conditions. In radical pathway, cleavage type (type I) photoinitiators, H-abstraction type photo-initiators, or thermal initiators can be used [68]. Although a wide range of olefinic compound can be used in this reaction, reaction rates considerably change depending on the reaction mechanism and substitutions around the C=C bond. Besides, selection of thiol compounds substantially affect the rate of the reaction depending on S-H bond strength and the cleavage pattern [69].

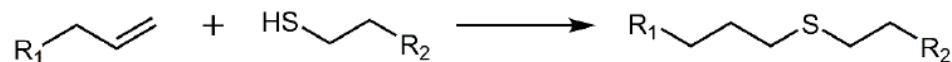


Figure 3: Thiol-ene click reaction mechanism.

Thiol-ene click reaction has been employed in the preparation of various well-defined polymeric structures. Iskin *et al.* gave a good example in their report on the synthesis of an ABC-type miktoarm star polymer through thiol-ene, ROP, and CuAAC reactions. Firstly, they synthesized a core compound with allyl, hydroxyl, and azide functional groups. Then, thiol end-capped PSt was prepared via ATRP and suitable chemical functionalization methods. Thiol-functional PSt and allylic compound were reacted in the presence of bis-(2,4,6-trimethylbenzoyl)-phenylphosphine oxide (BAPO), which is a cleavage type photoinitiator, yielding azide and hydroxyl functional PSt. Subsequently, the obtained PSt was employed in the ROP of ϵ -CL, giving PSt-*b*-PCL diblock copolymer with azide unit at the core. PSt-*b*-PCL was clicked with alkyne end-capped PEG, yielding PSt-PCL-PEG miktoarm star polymer [70].

Thiol-ene Michael addition reactions slightly differs from radical-mediated thiol-ene reactions. In these reactions, the abstraction of hydrogen with suitable bases produces an ion. Then, the thiolate anion is attached to an olefinic moiety (having an e-withdrawing unit) and then a carbanion intermediate is produced, which in turn takes a hydrogen from another thiol, yielding a new thiolate ion. The reaction goes on until all double bonds are used up [71].

Thiol-ene Michael reactions are very effective green conjugation techniques and used in many biochemical functionalization procedures and surface functionalization of biomaterials [72, 73]. Additionally, conjugation of polymers or of small molecules to polymers can be accomplished via nucleophile- and base-catalyzed thiol-ene Michael reactions even in very dilute systems without resulting in detrimental side products [74]. Yu *et al.* reported an elegant synthetic procedure for the preparation of sequence-ordered polymers via sequential thiol-ene Michael addition and radical mediated thiol-ene reactions in one-pot manner. The thiols which were produced by in situ ring-opening reactions of thiolactones readily reacted with electron-deficient methacrylate double bond of the allyl methacrylate (not the electron-rich allylic double bond) via thiol-ene Michael addition. Subsequent to the

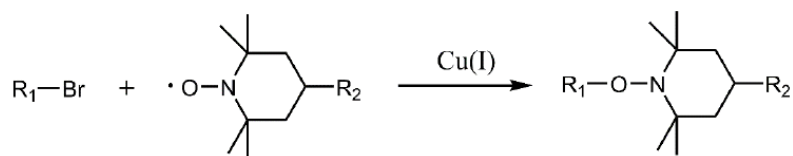


Figure 4: ATNRC click reaction procedure.

completion of the Michael addition and ring opening reactions, radical-mediated thiol-ene click reaction was performed under UV irradiation [75].

5. ATOM TRANSFER NITROXIDE RADICAL COUPLING (ATNRC) CLICK REACTIONS

Atom transfer nitroxide radical coupling is one of the strongest and the most selective ligation methods. Although it is only a ligation method, its mechanism resembles that of atom transfer radical polymerization and nitroso-mediated polymerization (Figure 4). Nitroxide radical cannot start polymerization by attacking olefinic bonds, instead it only couples with carbon-centered radicals at diffusion-controlled rates ($k_t \sim 10^8 \text{ L mol}^{-1} \text{ s}^{-1}$) [76]. Lin *et al* reported a very good example of ABC-type triblock linear copolymers via in situ CuAAC and ATNRC reactions. Azide end-capped PtBA, PSt with α -alkyne and ω -bromide functionalities, 2,2,6,6-tetramethylpiperidine 1-oxyl (TEMPO) end-capped PEG and PCL polymers were

prepared by living polymerization methods. PtBA-*b*-PSt-*b*-PEG and PtBA-*b*-PSt-*b*-PCL triblock copolymers were prepared via in situ CuAAC and ATNRC click reactions of corresponding homopolymers in the presence of CuBr/PMDETA catalyst system [77].

6. ALDEHYDE-AMINOXY CLICK REACTIONS

Aldehyde-aminoxy reactions are accepted as biorthogonal click reactions and include condensation reactions of aldehydes or ketones with α -effect amines (hydrazides or aminoxy reagents) [78]. This reaction method was used for the preparation and chemical modification of different biologically important macromolecular structures [79].

Wu *et al.* synthesized PNIPAM-PEG miktoarm star polymer via RAFT polymerization of N-isopropylacrylamide (NIPAM) and aldehyde-aminoxy conjugation method (Figure 5). PNIPAM macro-RAFT agent was prepared via RAFT polymerization of N-

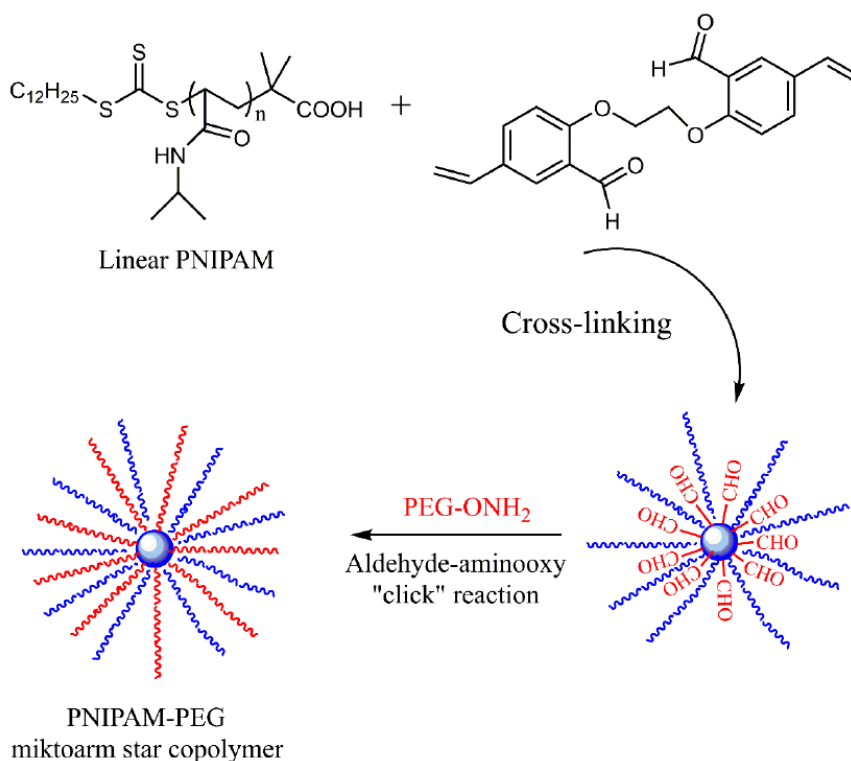


Figure 5: Synthesis of PNIPAM-PEG miktoarm star polymer via RAFT polymerization of poly(N-isopropylacrylamide) (PNIPAM) and aldehyde-aminoxy conjugation method.

isopropylacrylamide (NIPAM) using 2-(dodecylsulfanylthiocarbonylsulfanyl)-2-methylpropionic acid (DMPA) RAFT agent in the presence of 2,2'-azobisisobutyronitrile (AIBN). Then, it was used as the macroinitiator in the crosslinking reaction of the aldehyde functionalized divinyl compound. The obtained multiarm star polymer had a crosslinked core with aldehyde functional groups. Then, the aldehyde groups at the core were conjugated with aminoxy functional PEG, yielding miktoarm star with PNIPAM and PEG polymers chains [80].

7. CONCLUSION

Click chemistry techniques are very effective synthetic tools for the preparation of tailor-made elegant polymeric structures. Among them, Cu(I)-catalyzed azide-alkyne [3+2] dipolar cycloaddition reaction (CuAAC) was regarded as "the cream of the crop" due to its high tolerance to variety of functional units and near-perfect yields in very mild conditions. Whereas the use of transition metal catalyst is a major drawback limiting its use in biochemical applications. Since Diels-Alder, thiol-ene, and aldehyde-aminoxy click reaction methods do not involve metal catalyst species, they can be safely employed in the synthesis of macromolecules for biological and biochemical applications.

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